

OTPIP XXVIII

From a ToF-SIMS study on glass corrosion to self-healing surfaces

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Weinbergcampus in Halle/Saale



1992/2016
Foundation

330
Employees

Halle
(Main-)Site



IMWS/CAM
(Center for Applied Microstructure Diagnostics)

Fraunhofer IMWS

Who we are

1

We address the smallest dimension - with microstructure-based diagnostics and technology development for innovative materials, components and systems.

2

We develop solutions for functionality, reliability, safety, durability, sustainability and accelerated development of materials.

3

We increase the material efficiency and cost-effectiveness of materials and systems, thus helping to conserve resources.



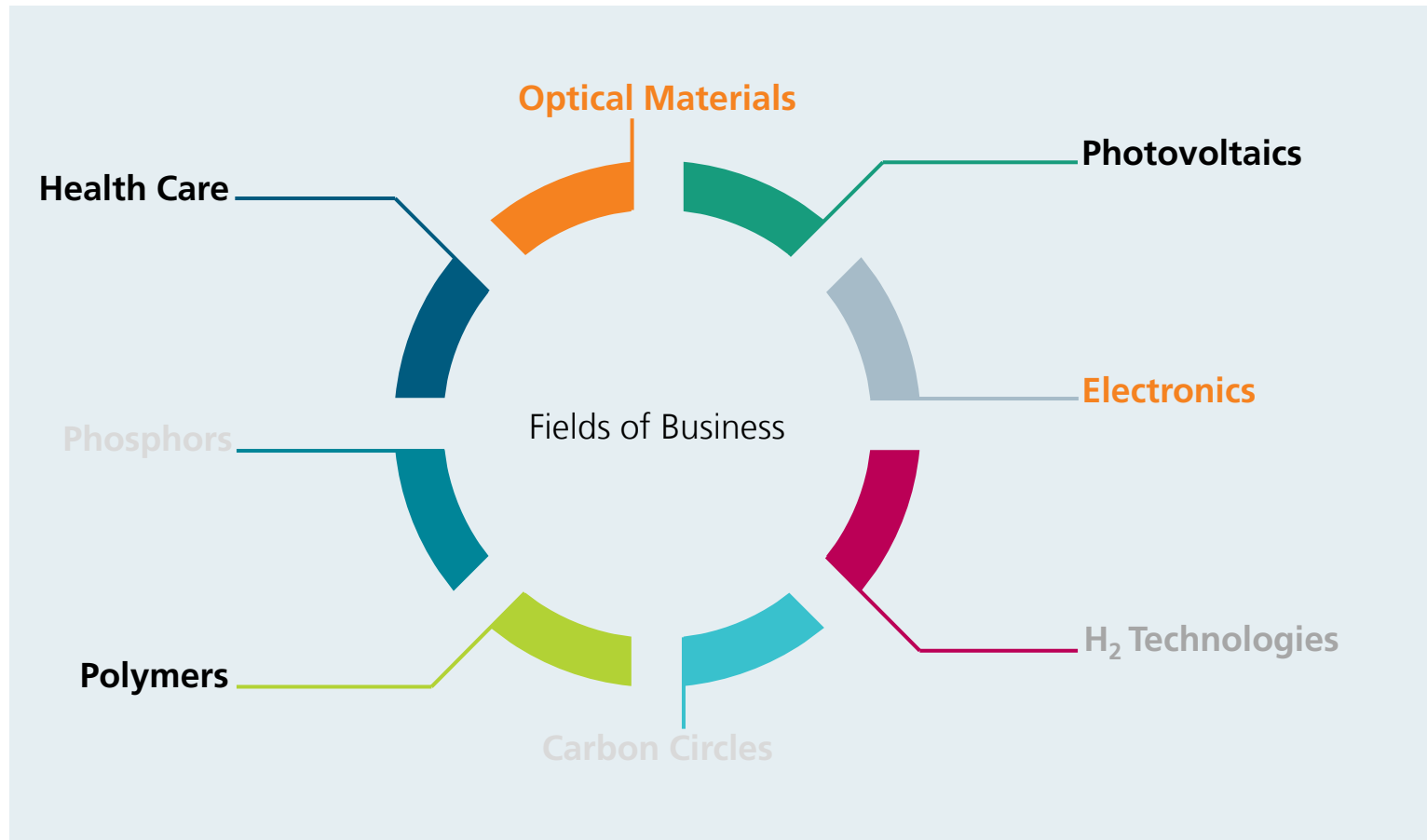
Materials Insights

From the microstructural material and component characteristics, statements can be made about the properties in the application case.

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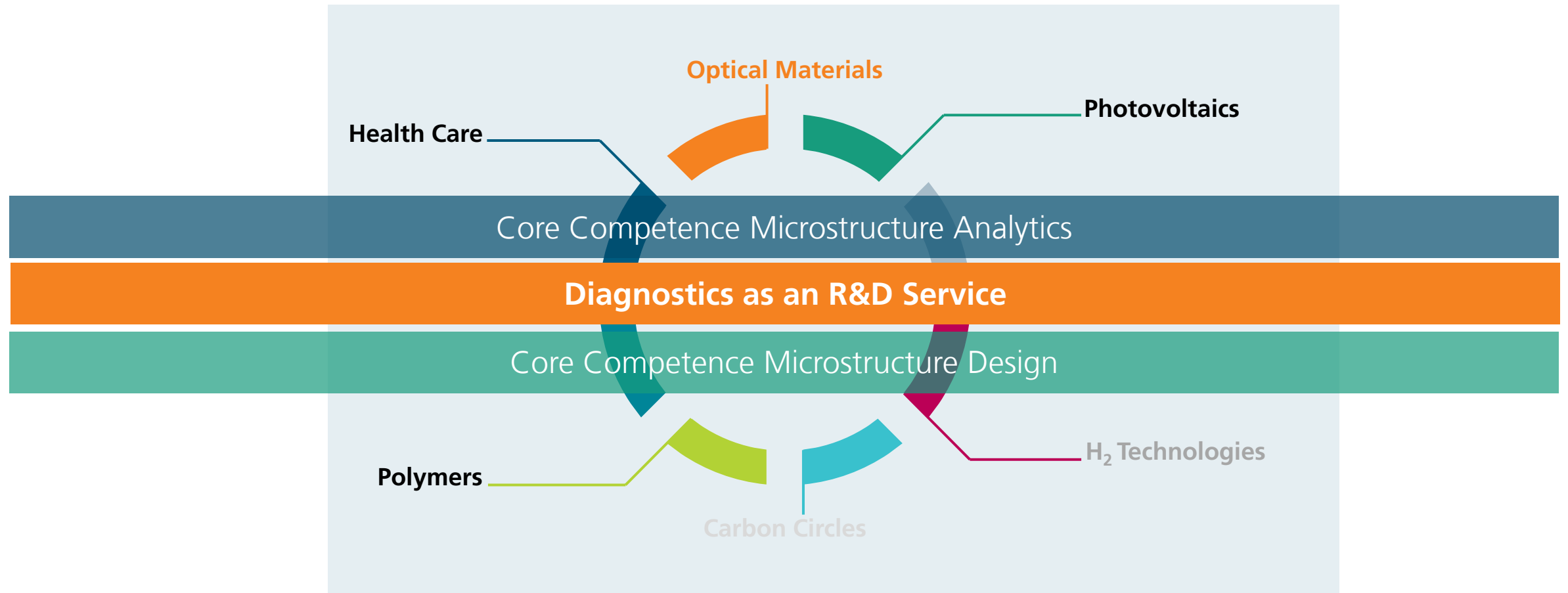
Fraunhofer IMWS

Methodically Oriented Portfolio Institute for Microstructure Diagnostics & Microstructure Design



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Business Unit «Optical Materials and Technologies»

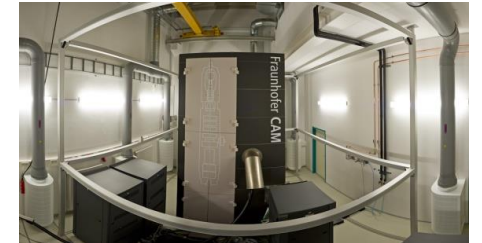
Service Offering Consists of Three Pillars

Analytics

1

Microstructure diagnostics, on industry demand

Including sample prep for optical materials
(substrates, thin films, pigments)



Technology
Development

2

Microstructure-based Materials Development

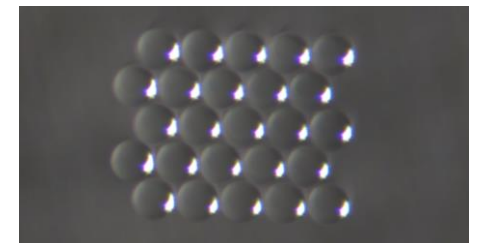
Glasses [seals, etc.] & glass ceramics



3

**Microstructure-based Optimization of
Micromachining**

using Lasers & ion beams



Corrosion Studies

Glass as a Material

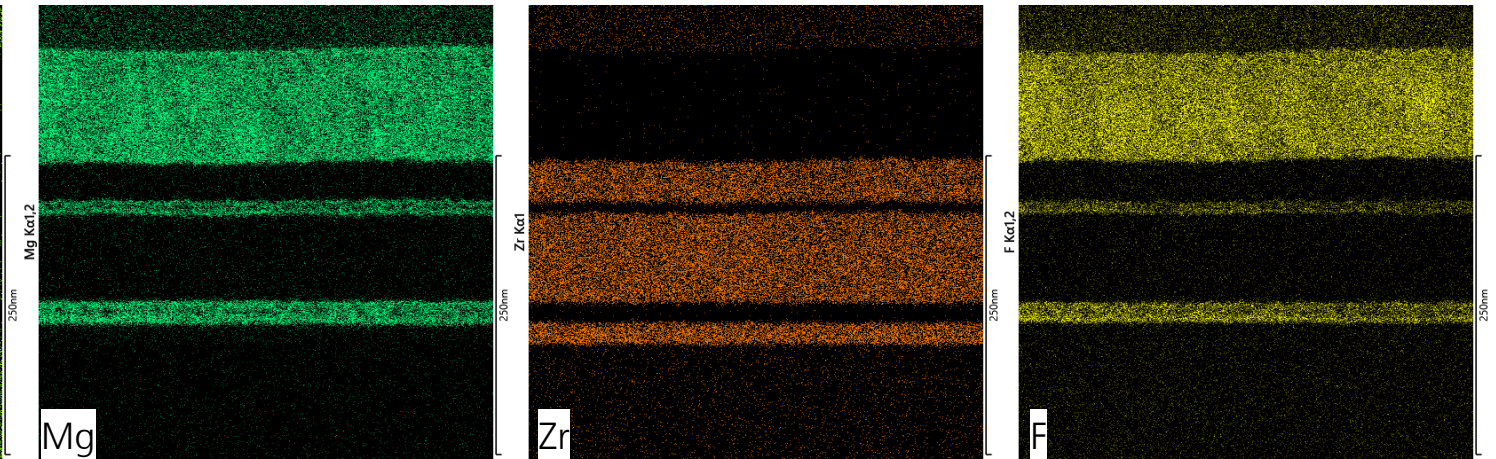
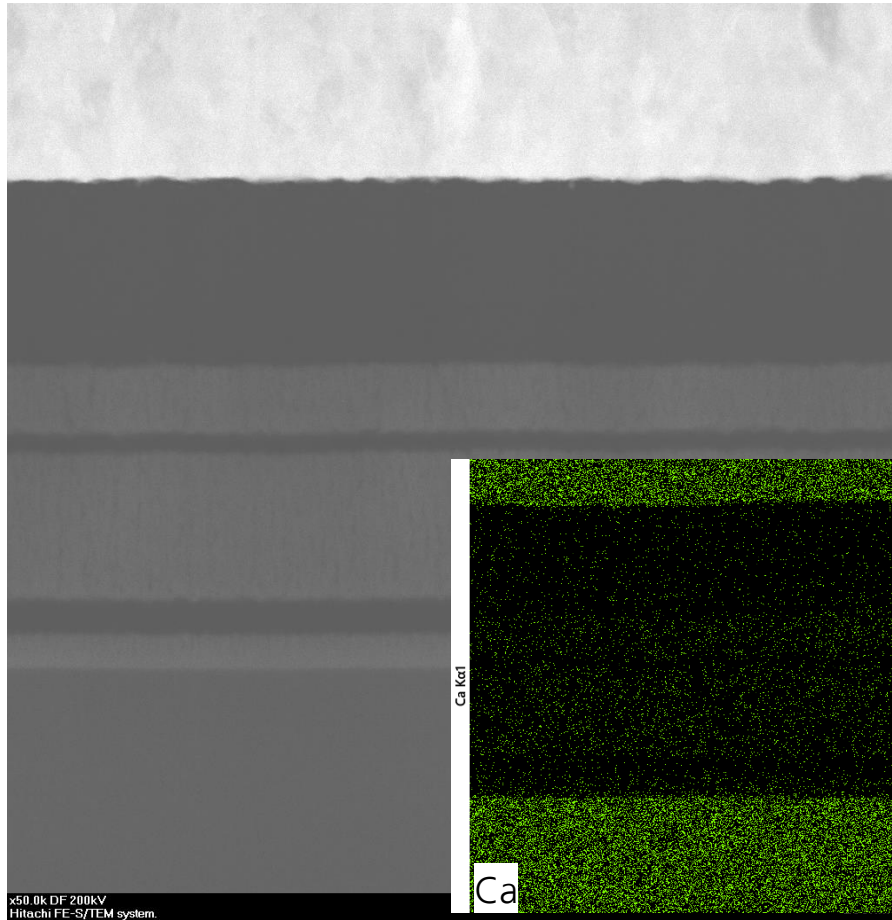


- Earliest traces of glass made by humans: > 5000 years
- Optical properties in visible range
- Great stability over time, but sensitive to stress
- Glass corrosion: depending on environmental conditions
- Industrial approach to increase durability:
 - increased hardness, especially for edges and corners
 - chemical resistance

Corrosion Studies

Glass Surface Functionalization

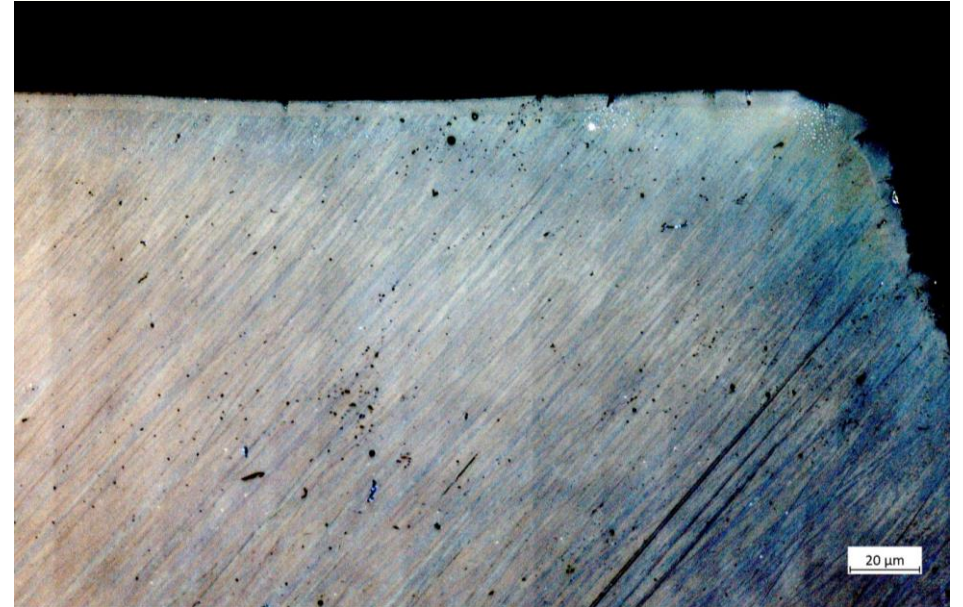
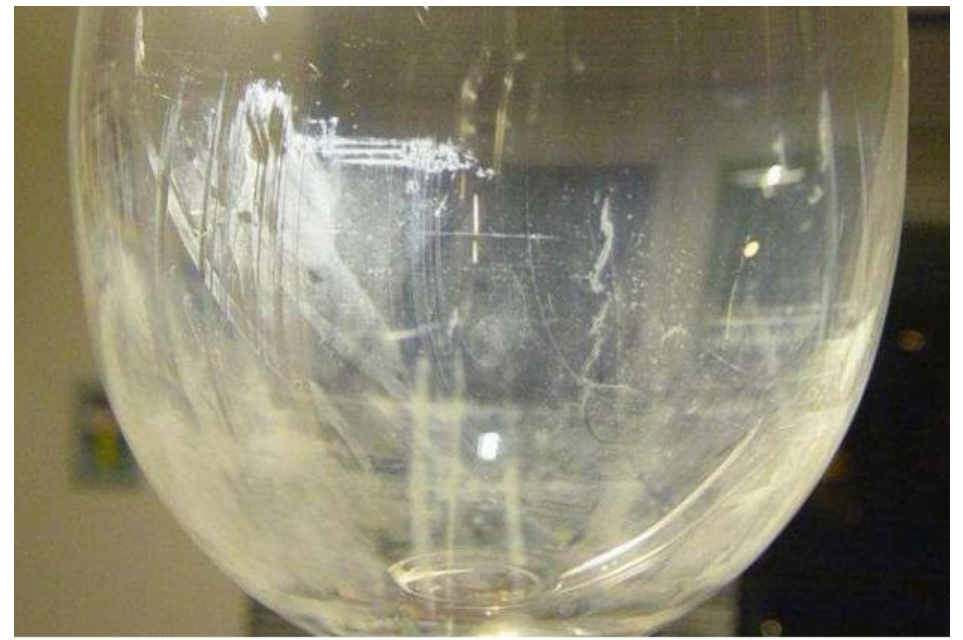
- Hardening of the glass surface:
 - by ion exchange processes
 - by tempering (heating and rapid cooling of the surface)
- Improving chemical stability and optical properties:
 - deposition of (multi-layered) coatings



Corrosion Studies

Glass Corrosion

- General knowledge: corrosion processes of inorganic glasses are well understood.
- Ion exchange processes at the glass surface lead to the formation of a gel-like layer, influencing the mechanical and optical properties of the glass.
- Processes are accelerated in combination with water or humidity in the environment.
- The glass stoichiometry influences the corrosion behavior
- But: most of this knowledge is phenomenological.
- The microstructural mechanisms are an actual research topic, and the improvements in surface and microstructural analysis promise further insights.



How it Started

ToF-SIMS Corrosion Studies

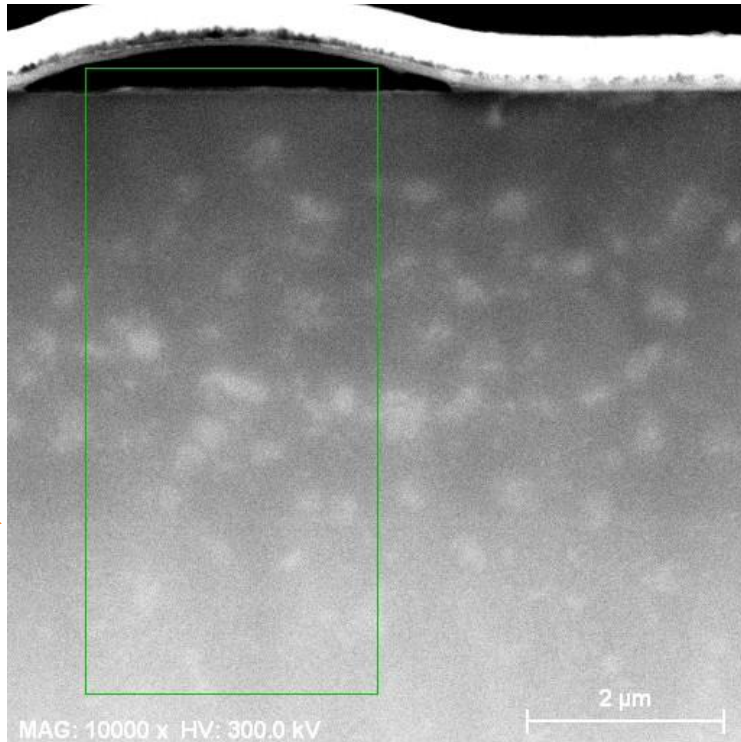


*85 SiO₂ - 15 Rb₂O glass
made at 1600 °C*

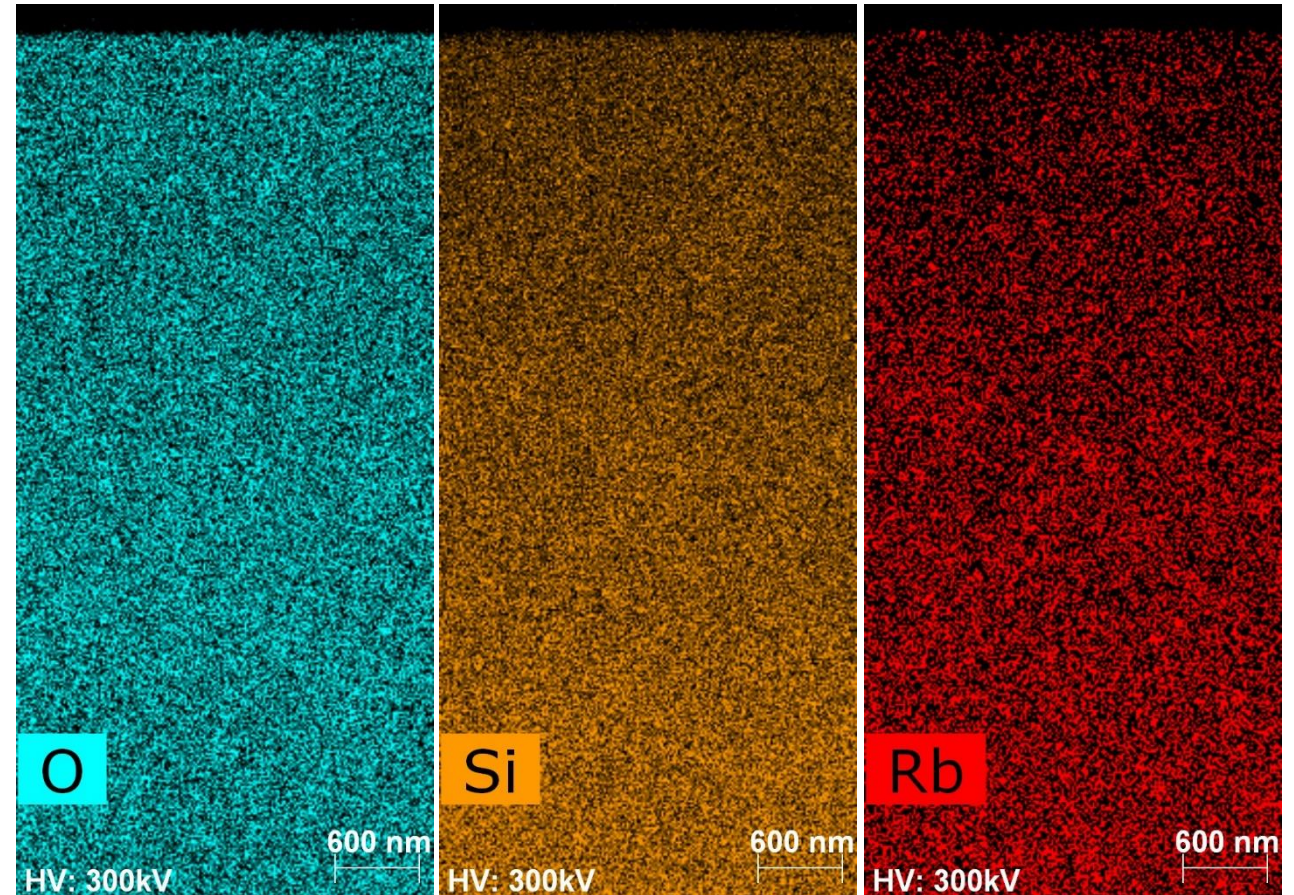
- Innovent and Fraunhofer IMWS wanted to improve the general knowledge about glass corrosion, in order to mitigate the process.
- Theoretical work at Innovent (Dr. Detlef Stock) indicated that superficial redistribution of elements/bonds might be key.
- IMWS suggested to have a look into the surface-near composition of a glass using time-of-flight SIMS.
- Due to its omnipresence on surfaces and very high mobility of sodium, it was decided to have a look at a quasi-binary, alkali silicate model glass $x \text{ SiO}_2 / (100-x) \text{ Rb}_2\text{O}$ (x=90, 85, and 80, mol%).
- A “thirsty glass” was produced by bubbling with dried Ar.
- ToF-SIMS proved CO₂- and water uptake by the $x \text{ SiO}_2 / (100-x) \text{ Rb}_2\text{O}$ glasses.

How it Started

STEM-EDX Corrosion Studies

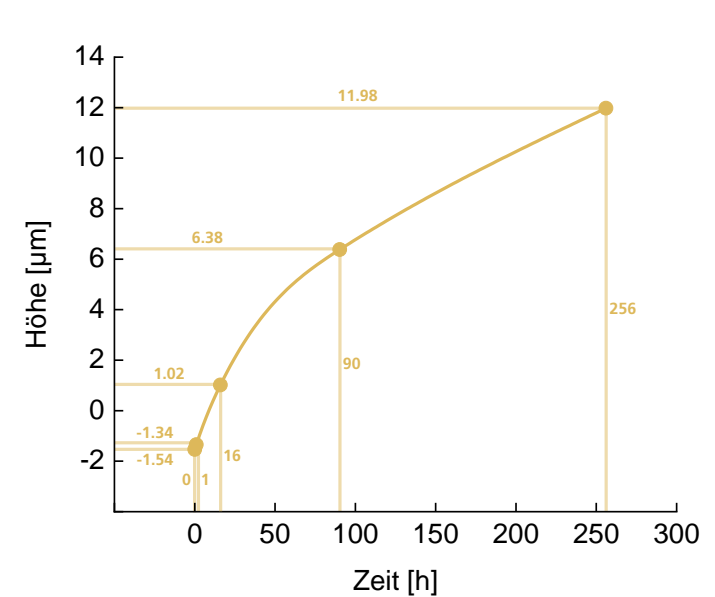
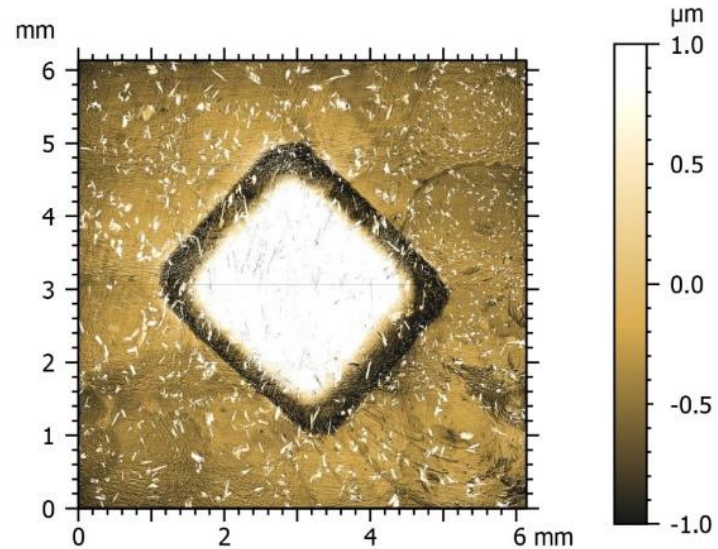


- TEM cross section also revealed some interface a few microns below the surface.



How it Started

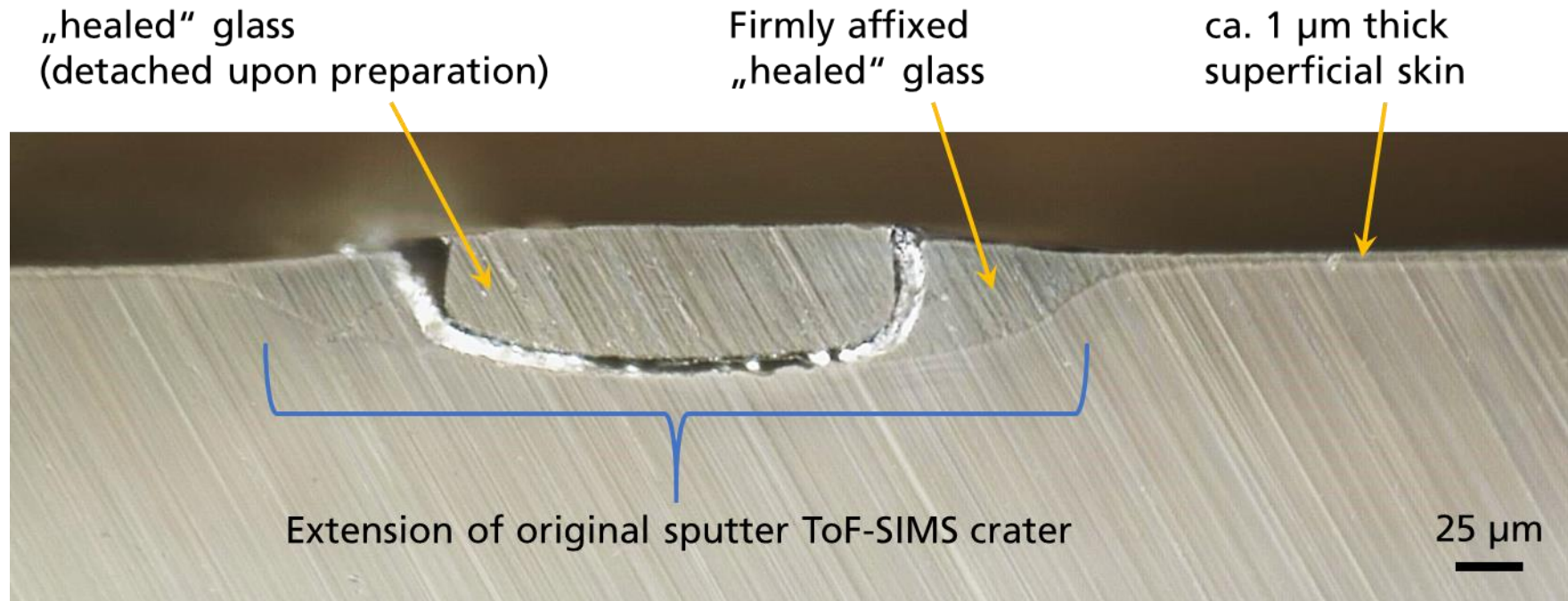
White Light Interferometry



- White light interferometry was used to measure the depth of the ToF-SIMS crater (crater depth vs. sputter time)
- The crater did not have a depth, but a *height* -> problem of different refractive indices?
- Mechanical profilometry and WLI after metal coating of the surface proved the effect.

How it started

Cross section



- Optical cross section of the surface skin and the recovered volume of a superficially harmed (ToF-SIMS depth profiling down to a depth of ca. 3 μm) 80 SiO₂ / 20 Rb₂O glass

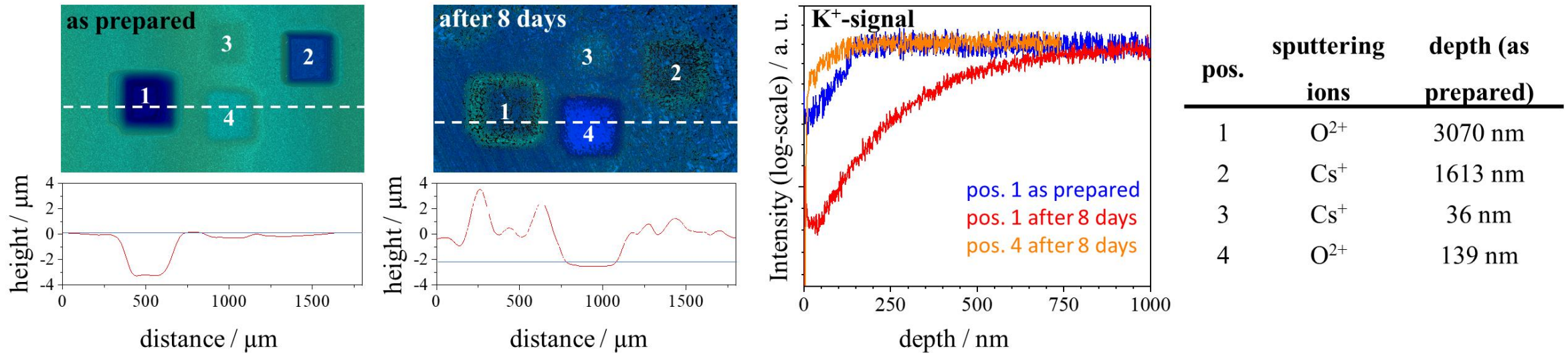
How it Started

Self Healing Glass Surface – possible explanation

- Rubidium silicate glasses are known to be highly hygroscopic.
- Therefore, contact with humidity was controlled in many ways:
 - Dry (bubbling with dried Argon) and standard melting of glass
 - Preparation of surfaces and cross-sections w and w/o water
 - Storage in exicators
- IR spectroscopy is a tool to measure the water content in glass.
- Glasses that are molten dry have the tendency to take up water afterwards.
- Upon this water uptake, the glass forms a skin that acts as a barrier against continued water uptake.
- Injuring this skin brings buried glass (that did not take up water so far) in contact with water vapor, resulting in self healing.
- **The effect is particularly pronounced for $\text{SiO}_2 - \text{Rb}_2\text{O}$ glasses, but likely to occur in other alkali silicate glasses (in a lesser amount) as well.**

How it Started

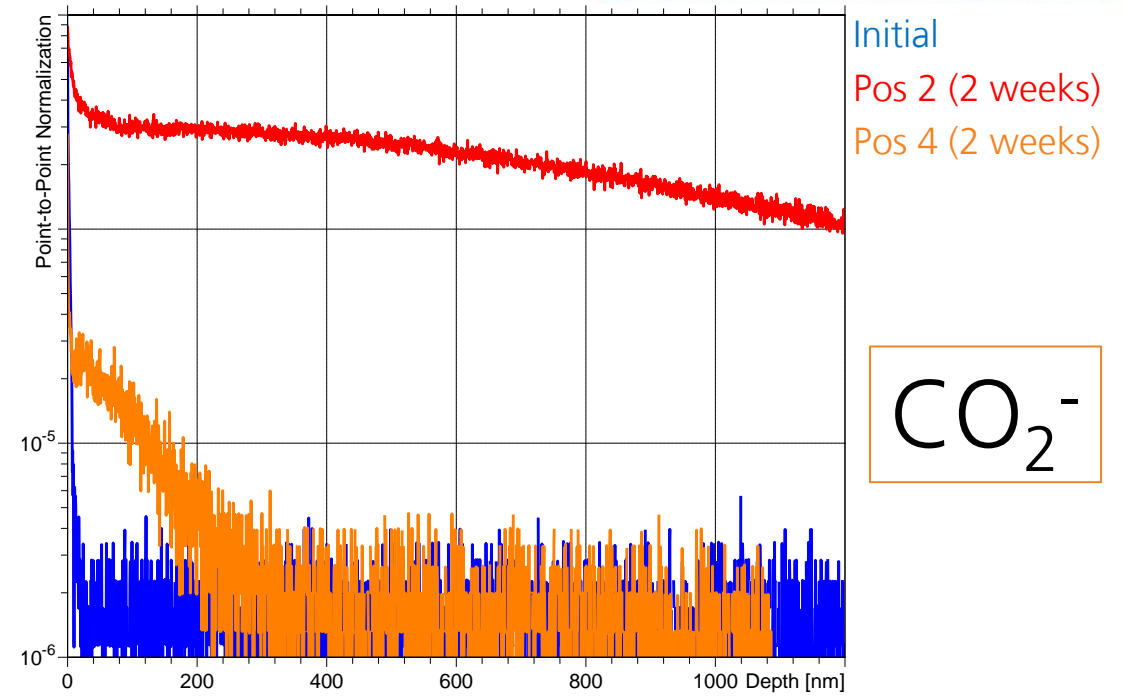
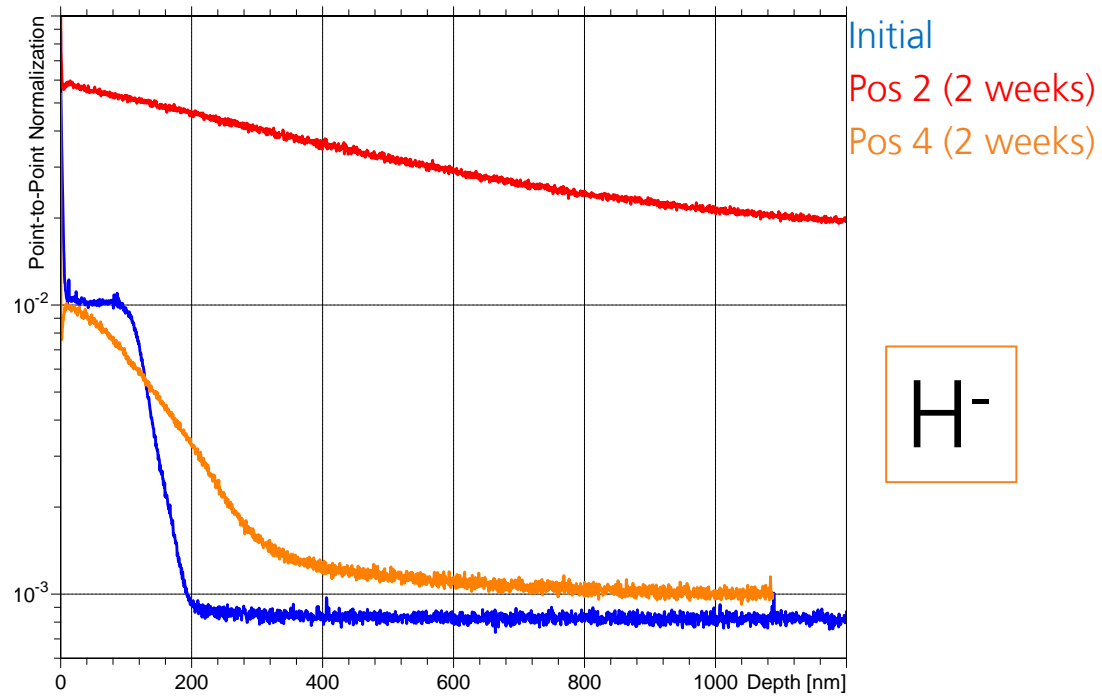
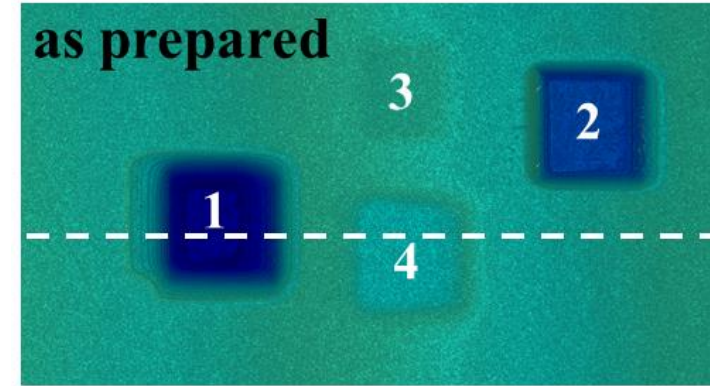
Proof of Principle



- Not just dry-molten 80 SiO₂ – 20 Rb₂O glass is prone to superficial changes after hurting the glass skin.
- Also less “exotic” glass compositions, like 75 SiO₂ / 25 K₂O glass, show self-healing (crater 1&2).
- At depth profiling position 4, oxygen ions were used to sputter a ca. 140 nm deep excavation. The depth of the excavation exactly corresponds to the thickness of the glass skin after gentle corrosion.
- Within 8 days, this sputter crater did not show degradation nor healing (K depletion, water take-up etc.).

How it Started

Proof of Principle



- Initial depth profiling of craters #2 & #4 compared to signal after 2 weeks: complete removal of surface layer leads to a pronounced water and CO2 uptake.

And Now?

Variation of Material Systems

1	1	1,008	
1	H		
	Wasserstoff	2,2	0,09
2	3	6,94	4
2	Li		
	Lithium	0,98	0,53
		11	22,990
3	Na		
	Natrium	0,93	0,97
		19	39,098
4	K		
	Kalium	0,82	0,86
		37	85,468
5	Rb		
	Rubidium	0,82	1,53
		55	132,91
6	Cs		
	Caesium	0,79	1,90
		87	223,03
7	Fr		
	Francium	0,7	?



- The properties of a glass are dominated by the stoichiometry of its oxide components. However, even at temperatures above 1000 °C, water is incorporated into the structure (OH-), typically between 300 ppm and 400 ppm.
- “Thirsty” glasses can be produced by bubbling (<10 ppm) with dried Ar.
- As Rb₂O-SiO₂ is quite exotic, other alkali-containing glasses have to be tested (Na, K).
- Variation of Rb₂O (or Na, K) content and water content.
- Necessary: characterization of produced glasses (DSC, IR...).

And Now?

Variation of Damage



- Ion species (O, Cs, Ar, Ga, Xe, Kr), energy and sputter depth.
 - Laser damage (ablation or local melting).
 - mechanical scratching or tribological testing.
- **Mechanism of the formation of new material to form a flat (healing) or convex (structuring) surface?**

Summary

- A very “thirsty” hygroscopic glass was produced at IMWS ($\text{Rb}_2\text{O-SiO}_2$), showing an unknown and unexplained behavior after the surface was “damaged” by ToF-SIMS depth profiling.
- The crater with a depth of a few μm “healed” itself in just a few days.
- The effect could be reproduced using Ar (ToF-SIMS) and Xe (plasma FIB) ions.
- A local removal of material leads to the formation or growth of new glass at the surface, not limited by the initial surface.
- Microstructural analysis (STEM-EDX) do not show measurable differences in the glass stoichiometry of the healed area, compared to undamaged regions.
- The stress induced by preparation and analysis of the samples did not show significant differences between the new grown glass and the bulk material, indicating comparable durability.
- The effect could successfully be translated to another hygroscopic alkali glass system ($\text{K}_2\text{O-SiO}_2$).

Outlook

- FhG-internal Discover project @IMWS.
- The effect is particularly pronounced for SiO₂ – Rb₂O glasses, but likely to occur in other alkali silicate glasses (in a lesser amount) and in SiO₂-Rb₂O glasses with lower R₂O-content as well.
- Changing the stoichiometry influences the growth of the healed surface. Thus, healing of scratches without “overgrowing” should be possible (smarter self healing).
- Optically active structures can be thought of by defined injury of a suitable glass. Upscaling?
- Role of CO₂ to be clarified.
- SiO₂ – Rb₂O glass coating might be an unconventional way to protect arbitrary glass surfaces from breaking (instead of imposing strain by ion exchange).

Contact

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Equipment

Microstructure Diagnostics

Preparation



Grinding
Polishing
Embedding
Sawing
Laser machining
Coating

SEM (4x) SEM/FIB (7x)



EDXS
WDXS
EBSD
EBAC & EBIC
Ga⁺- & Xe⁺-FIB
Nano probing

S/TEM (3x)



EDXS (0,8 & 2 sr)
EELS
Nano diffraction
aberration corrected
60 – 300 kV
„single e⁻ detection“

ToF-SIMS (2x)



Depth profiling
Gas cluster ion source
Cryo- & heating option
Polymer analytics

XPS

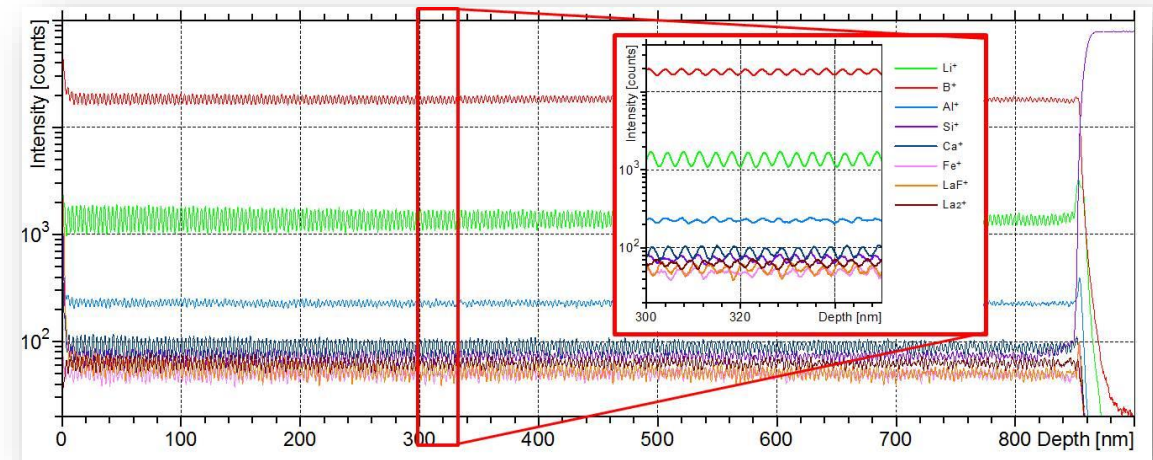
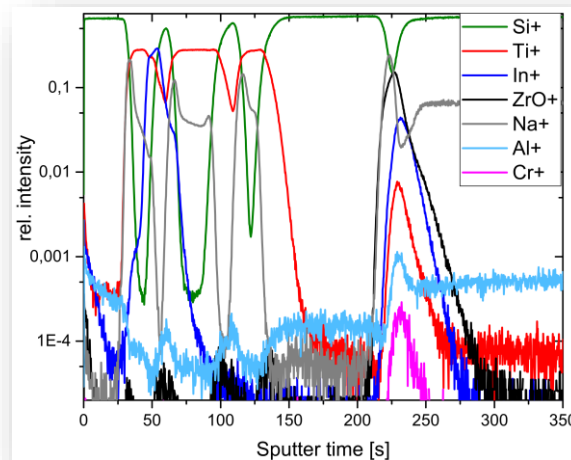


Chemical composition
Bonding stages

Optical Coatings

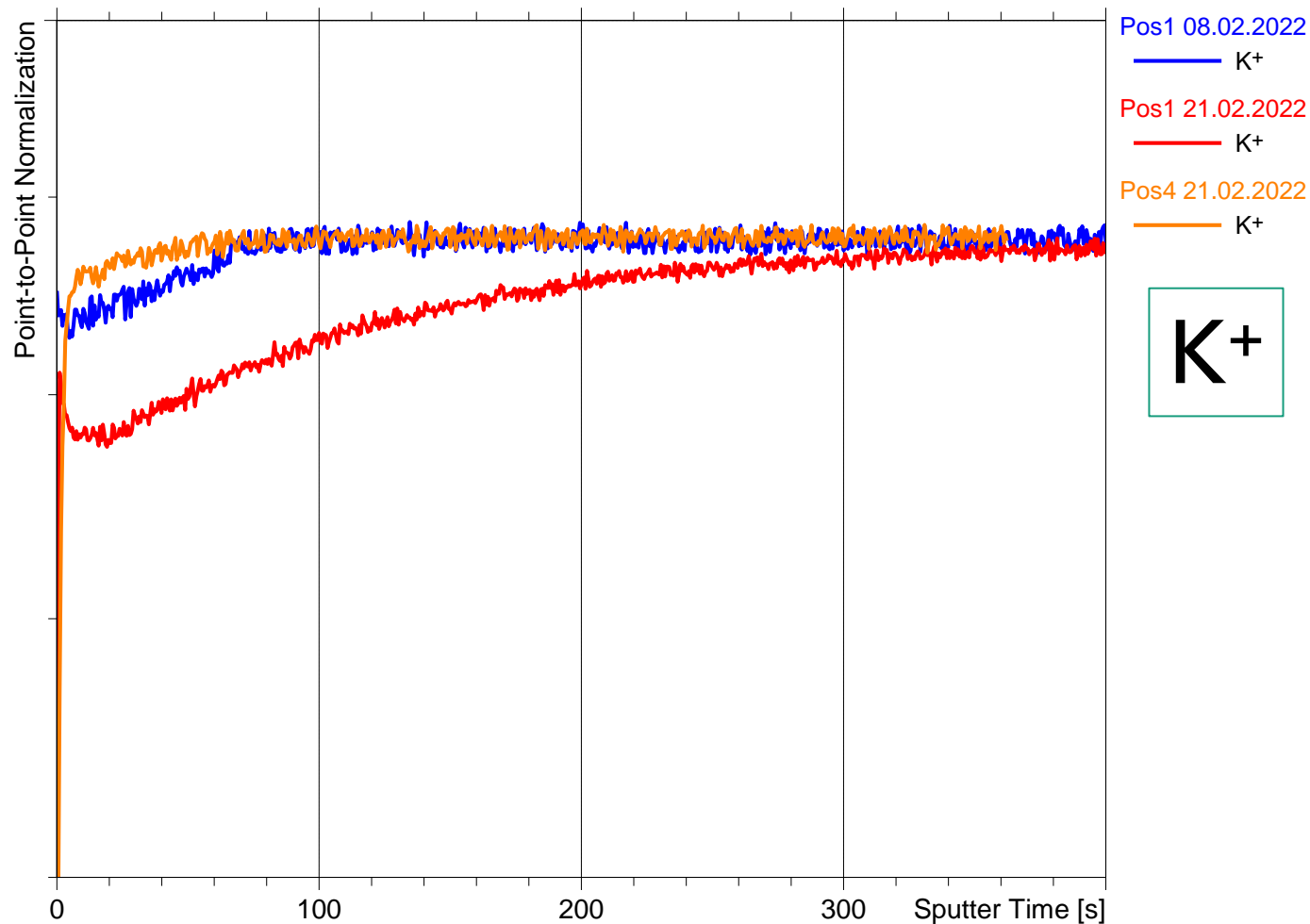
Typical Failures and Analysis Techniques: ToF-SIMS

- Time-of-flight secondary ion mass spectrometry: elemental and molecular composition of a surface with detection sensitivity in ppm-ppb region.
- Depth profiling via combination with different sputtering ion beams.
- Lateral resolution < 100 nm, depth resolution ~ nm.
- Detection of (interface) contaminations, interdiffusion and spurious elements.
- GCIB (organics) and cryo options available.

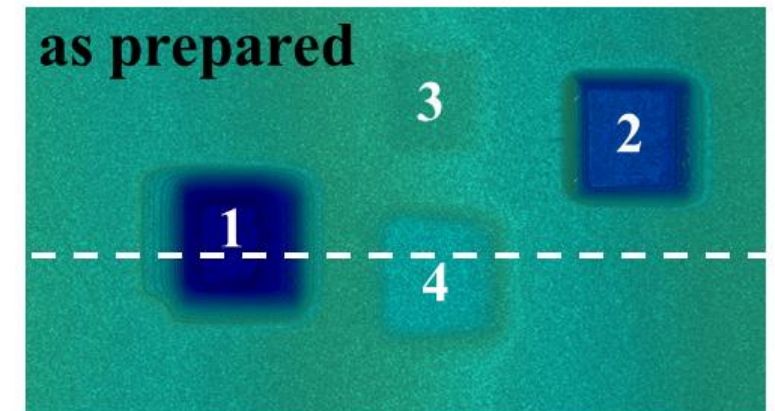


Depth Profiling, 0 – 400 s (≈ 820 nm)

Pos. 1 (pristine) = Reference vs. Pos. 1 (13 days later) vs. Pos. 4 (flat excavation, 13 days later)

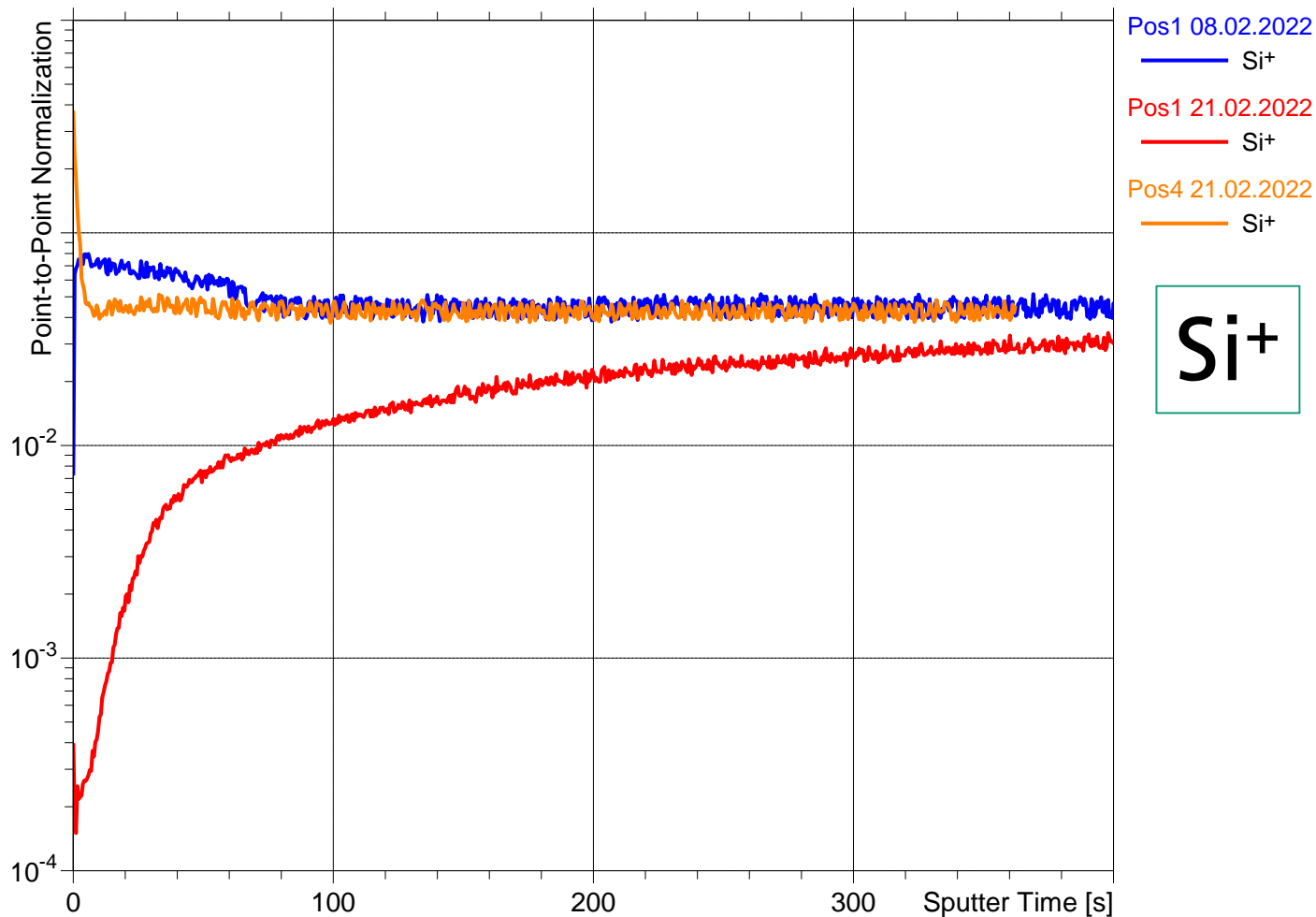


■ Alle signals shown here and herafter are normalised to the total intensity.

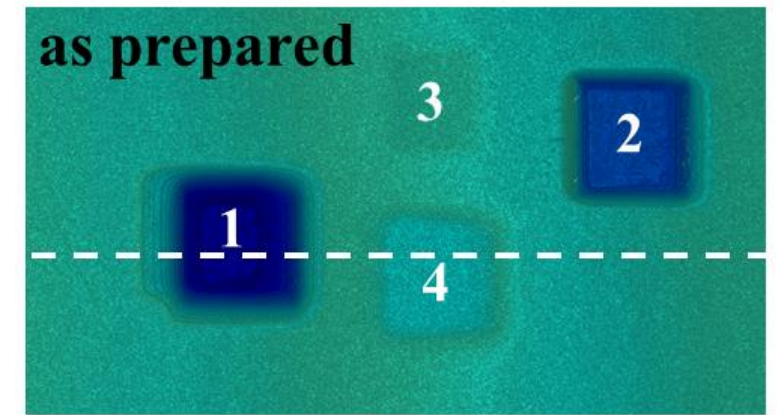


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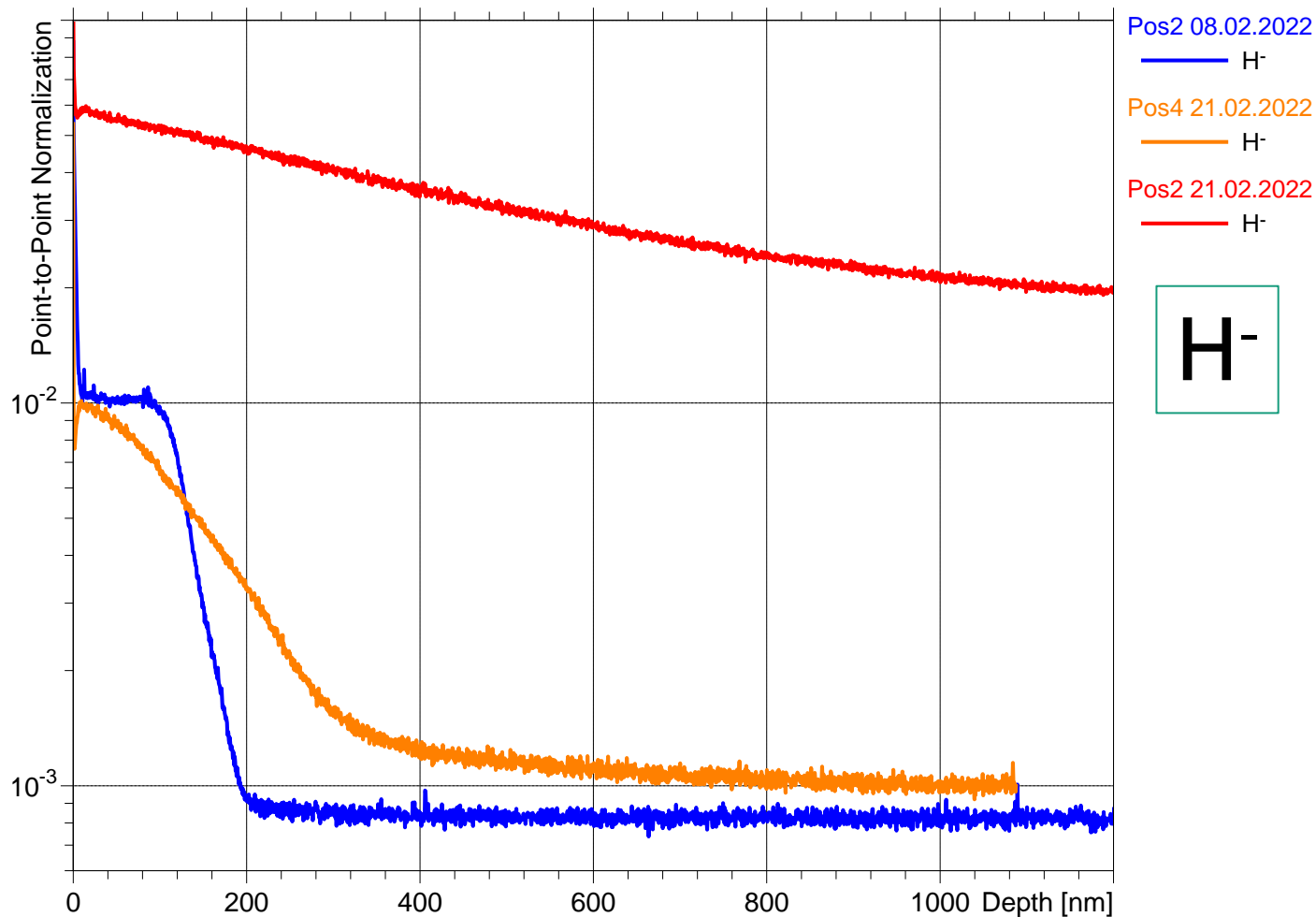


- Superficially, the smooth surface of position 4 features the highest silicon intensity.
- The latter steeply drops within the first ~ 20 nm below the surface in a constant level

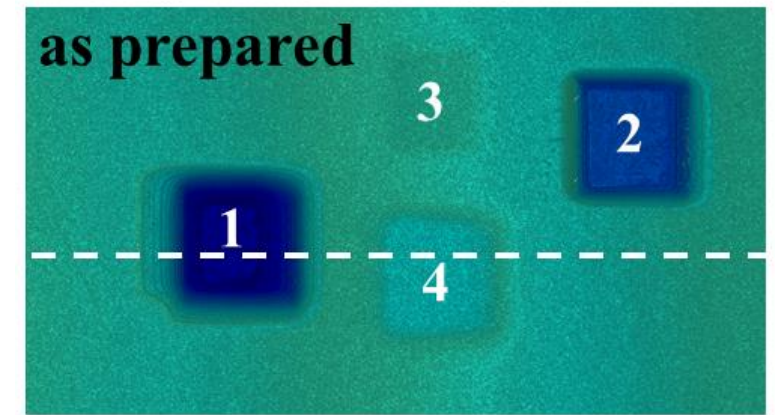


Depth Profiling, 0 – 1200 nm

Pos. 2 (pristine) = Reference vs. Pos. 2 (13 days later) vs. Pos. 4 (flat excavation, 13 days later)

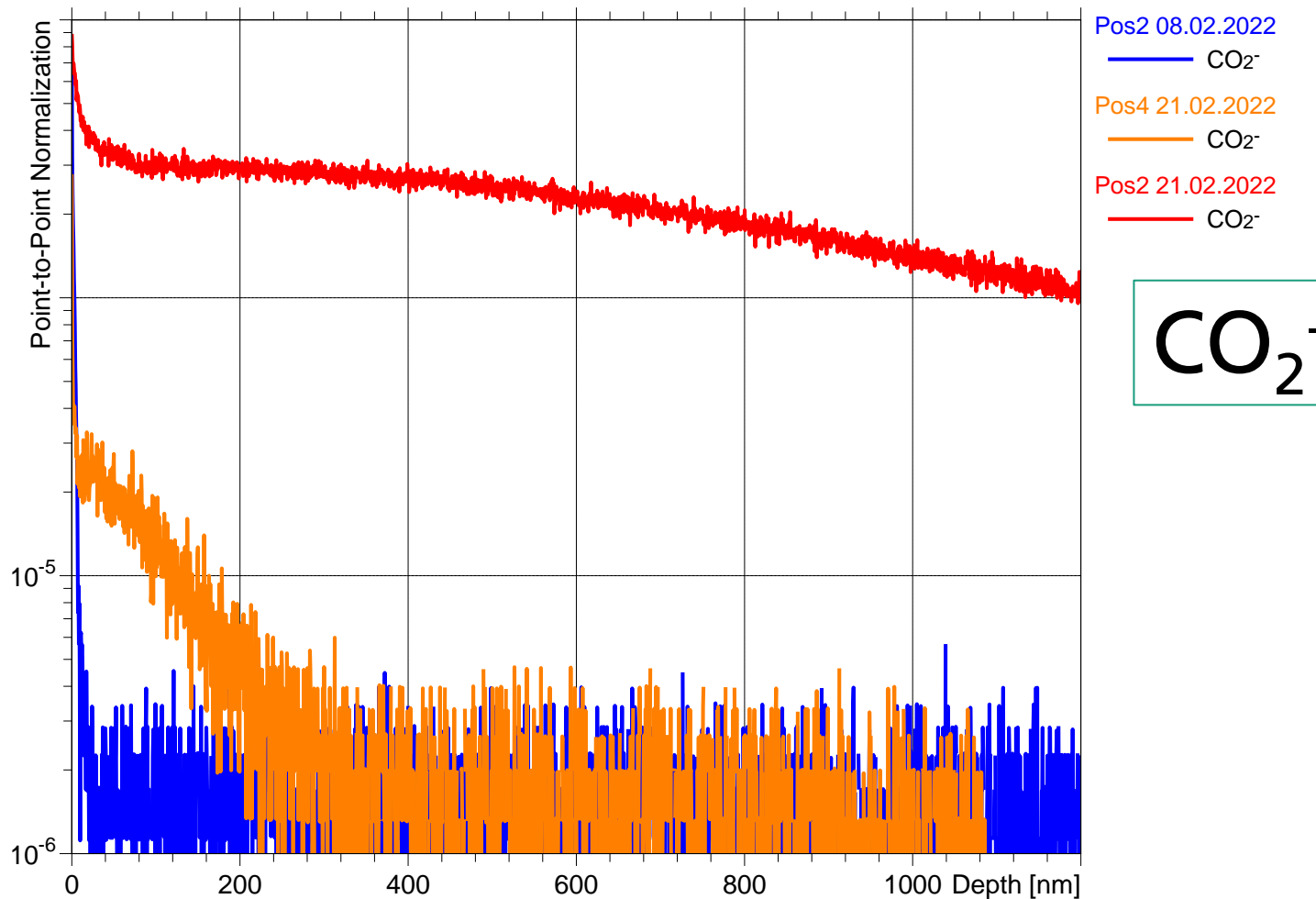


- In the pristine surface, the H signal is enriched within the topmost 10 nm, goes into a plateau going 150 to 200 nm below the surface and approaches than a one order of magnitude lower bulk level
- After ageing, the freshly formed surface features a much increase H content



Depth Profiling, 0 – 1200 nm

Pos. 2 (pristine) = Reference vs. Pos. 2 (13 days later) vs. Pos. 4 (flat excavation, 13 days later)



- Incorporation of CO₂ after removal of 1,6 μm by Cs⁺ sputtering
- Much reduced CO₂ incorporation after 140 nm O²⁺ sputtering.

